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Books A La Carte For Principles Of Chemistry: A Molecular Approach (2nd Edition)



Synopsis

This edition features the exact same content as the traditional book in a convenient, three-hole-punched, loose-leaf version. Books À la Carte also offer a great value—this format costs significantly less than a new textbook. Adapted from Nivaldo J. Tro's best-selling general chemistry book, *Principles of Chemistry: A Molecular Approach* focuses exclusively on the core concepts of general chemistry without sacrificing depth or relevance. Tro's unprecedented two- and three-column problem-solving approach is used throughout to give students sufficient practice in this fundamental skill. A unique integration of macroscopic, molecular, and symbolic illustrations help students to visualize the various dimensions of chemistry; and Tro's engaging writing style captures student's attention with relevant applications. The Second Edition offers a wealth of new and revised problems, approximately 50 new conceptual connections, and an updated art program throughout. This package contains: Books a la Carte for *Principles of Chemistry: A Molecular Approach*, Second Edition

Book Information

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Customer Reviews

This is a new textbook for general chemistry. Yes, there are quite a number of general chemistry texts and they all cover roughly the same general material. They differ more in how they organize the material, illustrate it, and their method of helping the student develop a solid understanding and command of this basic material. The student not only needs to learn the basic facts of chemistry, she needs to learn how to think about it so when she looks at the world, she can see and

understand the issues as they relate to this vital science. Dr. Tro begins this book by explaining how the text was developed and provides instructions for the student on how to make the most of the book. I liked this a great deal. He starts the book by providing what he thinks is the most important scientific idea in all of human knowledge: "the properties of matter are determined by the properties of molecules and atoms". From there he builds all the principles taught in general chemistry. He also defines chemistry as: "the science that seeks to understand the behavior of matter by studying the behavior of atoms and molecules". Each chapter is organized in a similar fashion and a great deal of emphasis is placed on making the illustrations demonstrate a principle - a process - rather than an isolated picture. Tro also has interesting articles of chemistry in various fields in different chapters and he often introduces ideas with practical issues that will likely matter to the student. For review, at the end of each chapter he goes over the key concepts taught, the key equations and relationships discussed, and the key skills that the student is expected to have learned. He provides exercises to test one's mastery of the material that the student can answer for himself.

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